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Highlights

► A concentrated shot of beetroot juice has a high total antioxidant capacity (TAC) and polyphenol (TP) content. ► TAC and TP are increased significantly ($P < 0.01$) following in vitro digestion. ► The beetroot shot (70 mL) provides a convenient method of increasing polyphenol consumption. ► Antioxidants from the beetroot shot are more bioaccessible than those from other vegetable juices.



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Short communications

A beetroot juice shot is a significant and convenient source of bioaccessible antioxidants

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ABSTRACT

The total antioxidant capacity (TAC) and total polyphenol (TP) content of a beetroot juice shot (70 mL) was examined following in vitro digestion. TAC was assessed using the ferric reducing antioxidant power (FRAP) assay and TP content was measured using the Folin Ciocalteu (FC) method (measured as gallic acid equivalents (GAE) before and after an in vitro digestion procedure with simulated gastric (GAS) and duodenal (DUO) phases. The beetroot shot had a high TAC ($697.9 \pm 1.6 \mu\text{mol}/70 \text{ mL}$) and TP content ($68.4 \pm 0.3 \text{ mg GAE}/70 \text{ mL}$). FRAP values increased approximately **3-fold** after GAS ($2361.2 \pm 20.9 \mu\text{mol}/70 \text{ mL}$) and remained high following DUO ($1740.3 \pm 21.1 \mu\text{mol}/70 \text{ mL}$). TP content increased **5-fold** following GAS ($341.6 \pm 4.8 \text{ mg GAE}/70 \text{ mL}$) and remained **3.3-fold** higher following DUO ($223.2 \pm 5.4 \text{ mg GAE}/70 \text{ mL}$). The beetroot shot delivers a high amount of bioaccessible antioxidants and may be a cost effective and convenient method of increasing antioxidant status.

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1. Introduction

The balance between the production of reactive oxygen species (ROS) and reactive nitrogen species (RNS), collectively termed 'RONS', and the protective mechanisms against them is considered important in preserving good health (Valko et al., 2007). Excess RONS production, termed 'oxidative stress', is implicated in the pathology of a number of serious disease states such as cardiovascular disease (Heitzer, Schlinzig, Krohn, Meinertz, & Munzel, 2001), cancer (Valko, Rhodes, Moncol,

Izakovic, & Mazur, 2006), and neurological decline (Jomova, Vondrakova, Lawson, & Valko, 2010) and can affect cell signalling (Hensley, Robinson, Gabbita, Salsman, & Floyd, 2000). Consumption of natural produce which is rich in antioxidant compounds may help to redress the balance between RONS production and endogenous protection when the body is under oxidative stress. A number of biological RONS such as superoxide anion (O_2^-), hydroxyl radical (OH^\bullet), nitric oxide (NO) and the peroxy radical (ONOO^\bullet) have been shown to mediate redox-dependant signalling pathways (Hensley et al., 2000).

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Abbreviations: FRAP, ferric reducing antioxidant power; FC, Folin Ciocalteu; GAS, gastric phase of in vitro digestion; DUO, duodenal phase of in vitro digestion; RONS, reactive oxygen and nitrogen species; GAE, gallic acid equivalents; TAC, total antioxidant capacity; TP, total polyphenols

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63 Antioxidants, both endogenous and exogenous are responsible
64 for regulating these actions by 'turning off' this pathway when
65 it is not beneficial.

66 Additionally, RONS can cause damage to lipid membranes
67 (through lipid peroxidation), proteins (through S-nitrosyla-
68 tion) and DNA (through nucleotide oxidation). Antioxidants
69 can directly inhibit the damage caused by these proliferative
70 reactions by donating the necessary electron to stabilise
71 RONS without becoming reactive in turn (Valiko et al., 2007).
72 In some disease states (e.g., cancer), where the production
73 of endogenous antioxidants including superoxide dismutase
74 (SOD) is disrupted, exogenous antioxidant consumption
75 may become more important in oxidative stress regulation
76 (Valiko, Izakovic, Mazur, Rhodes, & Telser, 2004). Thus the bal-
77 ance between the required RONS production for normal cell
78 signalling, and undesirable excess production leading to sub-
79 strate damage, is critical.

80 Beetroot juice contains a high level of biologically accessi-
81 ble antioxidants (Wootton-Beard, Moran, & Ryan, 2011) as
82 well as many other health promoting compounds such as
83 potassium, magnesium, folic acid, iron, zinc, calcium, phos-
84 phorus, sodium, niacin, biotin, B6 and soluble fibre. Addition-
85 ally, drinking beetroot juice provides a more convenient
86 alternative to consuming the whole vegetable. The specific
87 interest in beetroot juice has arisen because it is a rich source
88 of a number of polyphenolic compounds (Kaur & Kapoor,
89 2002; Pitalua, Jimenez, Vernon-Carter, & Beristain, 2010).

90 Beetroot predominately contains pigments called beta-
91 lains, a class of betalamic acid derivatives which are com-
92 posed of betacyanins and betaxanthins (Pitalua et al., 2010),
93 and a number of phenolic compounds. The betalain and phe-
94 nolic composition of red beetroot has been studied in detail
95 by Kujala, Loponen, Klika, and Pihlaja (2000) and Kujala, Vien-
96 ola, Klika, Loponen, and Pihlaja (2002). The first of these stud-
97 ies measured the effects of cold storage on the two main
98 betacyanins; betanin and isobetanin, and the ferulic acid es-
99 ter; β -D-fructofuranosyl- α -D-(6-O-(E)-feruloyl)glucopyranoside
100 (Kujala et al., 2000). The structures of these compounds are
101 shown in Fig. 1. A further study by this research group iso-
102 lated a number of other compounds in four beetroot cultivars
103 using high-performance liquid chromatography- electrospray
104 ionisation-mass spectrometry (HPLC-ESI-MS); Table 1 shows
105 the predominant betalain and phenolic constituents. Betanin
106 was shown to be the most abundant betalain in the flesh and
107 peel of each cultivar whilst 5,5',6,6'-tetrahydroxy-3,3'-biindol-
108 yl, feruloylglucose and β -D-fructofuranosyl- α -D-(6-O-(E)-fer-
109 uloyl)glucopyranoside) were all present in higher quantities
110 in the peel compared to the flesh. Vulgaxanthin I (shown in
111 Fig. 1) and II were the predominant betaxanthins (Kujala
112 et al., 2002). There are also other varieties of beetroot where
113 the ratio of betacyanins to betaxanthins changes which is
114 reflected in the colour of the root. Betanin, together with its
115 aglycone betanidin have been independently shown to have
116 high antioxidant activity (Butera et al., 2002; Tesoriere, Butera,

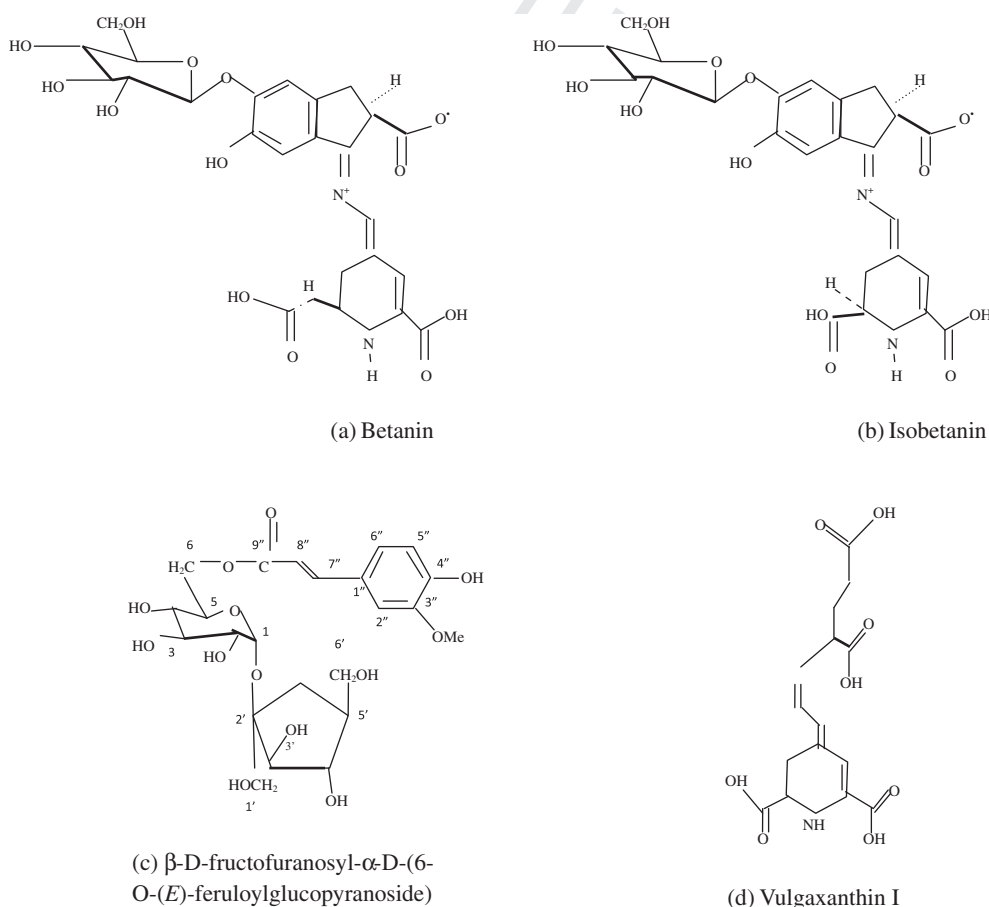


Fig. 1 – Structures of Betanin, Isobetanin, β -D-fructofuranosyl- α -D-(6-O-(E)-feruloyl)glucopyranoside) and Vulgaxanthin I.

Table 1 – Reported phytochemical composition of red beetroot, adapted from Kujala et al. (2002).

Classification	Compound
Betalains	
Betaxanthins	Vulgaxanthin I Vulgaxanthin II
Betacyanins	Betanin Isobetanin
Phenolics	
Ferulic acid conjugates	5,5',6,6'-tetrahydroxy-3,3'-biindolyl Feruloylglucose β -d-fructofuranosyl- α -d-(6-O-(E)-feruloylglucopyranoside)
Phenolic amides	N-trans-Feruloyltyramine N-trans-Feruloylhomovanillylamine
Flavonoids	Betagarin Betavulgarin Cochliophilin A Dihydroisorhamnetin

117 Pintaudi, Allegra, & Livrea, 2004). Beetroot also contains smaller
118 amounts of other compounds, such as carotenoids and
119 ascorbic acid which may further increase its total antioxidant
120 capacity.

121 Convenience is considered a considerable marketing tool
122 in the food industry (Drewnowski & Darmon, 2005), and conscientious
123 food manufacturers are looking for ways to make
124 healthy food, particularly fruit and vegetables, more convenient
125 to consume in order that public health may be
126 improved. There has been a significant increase in the number
127 of fruit based beverages which are available in UK commercial
128 outlets over the past decade and they have become
129 an important method of fruit intake, particularly for children.
130 By examining the shelves in local supermarkets it is clear to
131 see that vegetable juices are beginning to expand in the
132 same way. Fruit and vegetable blends in particular appear to
133 have become more popular, given the emergence of brands
134 such as V8 (Campbell Foods, Belgium), Cawston Press
135 (Wokingham, UK) and Sunraysia (Sunraysia UK Ltd, London,
136 UK). Vegetable juices are considered to be a viable method
137 of bridging the gap between actual and observed vegetable intake
138 (Shenoy et al., 2010). Recent research in our laboratory
139 has aimed to identify the antioxidant content of a number
140 of these juices in order that their contribution to dietary intake
141 can be quantified (Wootton-Beard et al., 2011). Following
142 this work, beetroot juice was found to be a particularly rich
143 source of antioxidants and polyphenols as measured using
144 a variety of biochemical methods before and after in vitro
145 digestion. However, beetroot juice consumption may not be
146 as popular as other fruit and vegetable juices such as tomato,
147 carrot, apple or mango, perhaps due to perceived issues of
148 taste, texture and urinary colouration. In reality beetroot juice
149 has a relatively pleasant taste in comparison with other vegetable
150 juices due to its relatively high sugar content (Thakur &
151 Das Gupta, 2006). A small increase in the habitual consumption
152 of polyphenol-rich beverages such as beetroot juice

153 may have a significant positive effect on public health. Recently,
154 a number of more convenient vegetable juice products
155 have been created to try to increase consumption, including
156 popular brands of mixed vegetable juice which are available
157 in cans and small bottles (V8, Campbell Foods, Belgium).
158 Additionally a beetroot juice 'shot' (70 mL) has been developed
159 (James White Drinks, Ipswich, UK). This particular product
160 offers an opportunity for beetroot juice to be consumed
161 with ease and convenience by the general public and may
162 contribute positively toward increasing consumption of polyphenol
163 rich produce.

2. Materials and methods 164

2.1. Materials and sample preparation 165

166 All chemicals were of analytical grade and were purchased
167 from Sigma Aldrich (Poole, UK). Beetroot juice shots were
168 provided by James White Drinks Ltd, Ipswich, UK. The beetroot
169 shot was analysed in its original form. In all experiments,
170 the beverages were prepared using a standard protocol.
171 Amber bottles were used throughout to prevent the photode-
172 composition of antioxidants and efforts were made to exclude
173 oxygen contact with the samples. All experiments
174 were carried out on a minimum of three separate occasions.
175 Samples were analysed in triplicate for each experiment.

2.2. Determination of total antioxidant capacity and total polyphenols 176

177 The total antioxidant capacity of the samples was determined
178 using a modification of the FRAP assay of Benzie and Strain
179 (1996). Briefly FRAP reagent was prepared from 300 mM acetate
180 and glacial acetic acid buffer (pH 3.6), 20 mM ferric chloride
181 and 10 mM 4,6-tripyridyl-s-triazine (TPTZ) made up in
182 40 mM HCl. All three solutions were mixed together in the
183 ratio 10:1:1. The FRAP assay was performed by warming
184 1 mL of dH₂O to 37 °C before adding 25 μ L of sample and
185 1 mL of reagent and incubating at 37 °C for 4 min. Absorbance
186 at 593 nm was determined relative to a reagent blank also
187 incubated at 37 °C. The total antioxidant capacity of samples
188 was determined against a standard of known FRAP value, ferrous
189 sulphate (1000 μ M). Total polyphenol content was determined
190 using the Folin Ciocalteu method (Singleton, Orthofer,
191 & Lamuela-Raventos, 1999). The beetroot juice aliquot
192 (0.2 mL) was added to 1.5 mL of freshly prepared Folin Ciocalteu
193 reagent (1:10, v/v, with water). The mixture was allowed to
194 equilibrate for 5 min and then mixed with 1.5 mL of 60 g/L
195 sodium carbonate solution. After incubation at room temperature
196 for 90 min, the absorbance of the mixture was read at
197 725 nm using the respective solvent as blank. The results
198 were expressed as mg of gallic acid equivalents (mg GAE).
199

2.2.1. In vitro digestion procedure 200

201 Analysis was repeated following in vitro digestion. The
202 in vitro digestion model was adapted from Ryan, O'Connell,
203 O'Sullivan, Aherne, and O'Brien (2008). Beetroot juice samples
204 were transferred to clean amber bottles and mixed with
205 Hanks' balanced salt solution (with NaCO₃, without phenol

Table 2 – Total antioxidant capacity and total polyphenol content of the beetroot juice shot and other beverages.

Product	FRAP (μmol)		FRAP post digestion (μmol)		TP (mg GAE)		TP post digestion (mg GAE)	
	Per L	Per serving	Per L	Per serving	Per L	Per serving	Per L	Per serving
James White Beetroot Shot	9971 \pm 22	697.9 \pm 1.6	24,862 \pm 300 [*]	1740.3 \pm 21.1 [*]	977.2 \pm 5.2	68.4 \pm 0.3	3189.1 \pm 77.3 [*]	223.2 \pm 5.4 [*]
James White Beetroot Juice ^a	8354 \pm 84	584.8 \pm 5.9	12,152 \pm 336	850.6 \pm 23.5	1450.3 \pm 42.1	101.5 \pm 2.9	1527.1 \pm 18.0	106.9 \pm 1.3
V8 Vegetable Juice ^a	2573 \pm 48	180.1 \pm 3.4	3488 \pm 223	244.2 \pm 15.6	617.8 \pm 15.6	43.2 \pm 1.1	1053.9 \pm 6.4	74.7 \pm 0.5
V8 Fruit and Vegetable Juice ^a	2060 \pm 49	144.8 \pm 3.4	4986 \pm 202	349.0 \pm 10.3	537.8 \pm 17.9	37.6 \pm 1.3	1097.4 \pm 7.5	76.8 \pm 0.5
Del Monte Tomato Juice ^a	2202 \pm 39	154.1 \pm 2.8	3071 \pm 50	215.0 \pm 3.6	695.3 \pm 19.3	48.7 \pm 1.4	1120.1 \pm 12.2	78.4 \pm 0.9
Eden Organic Carrot Juice ^a	1516 \pm 11	107.3 \pm 0.78	2731 \pm 27	191.9 \pm 1.5	473.7 \pm 7.7	33.2 \pm 0.5	1022.2 \pm 4.8	71.6 \pm 0.3

* Significantly increased compared to the juice prior to digestion, $P < 0.01$. FRAP = Ferric reducing antioxidant power. TP = Total polyphenols using Folin Ciocalteu method. GAE = Gallic acid equivalents. One serving represents a volume of 70 mL, equal to the total volume of the beetroot juice shot.

^a Data adapted from Wootton-Beard et al. (2011), values are given for comparative purposes and are not included in the statistical analysis for this experiment. Results are expressed as mean \pm SEM of three experiments each performed in triplicate. 'Post digestion' refers to the levels following the duodenal phase of the in vitro digestion procedure (Section 2.3).

red, sterile filtered, Sigma Aldrich, Poole, UK) to create a final volume of 20 mL. The samples were acidified to pH 2.0 with 1 mL of a porcine pepsin preparation (0.04 g pepsin in 1 mL 0.1 M HCl) and incubated at 37 °C in a shaking water bath at 95 rpm for 1 h. After gastric digestion, a 500 μL aliquot of each sample was stored at -20 °C. The pH was then increased to 5.3 with 0.9 M sodium bicarbonate followed by the addition of 200 μL of bile salts glycodeoxycholate (0.04 g in 1 mL saline), taurodeoxycholate (0.025 g in 1 mL saline), taurocholate (0.04 g in 1 mL saline) and 100 μL of pancreatin (0.04 g in 500 μL saline). The pH of each sample was increased to 7.4 with 1 M NaOH. Samples were incubated in a shaking water bath (95 rpm) at 37 °C for 2 h to complete the intestinal phase of the in vitro digestion process. After the intestinal phase, a 500 μL aliquot of each sample was stored at -20 °C, samples were analysed within 48 h.

2.3. Data analysis

All data are presented as means (\pm SEM) of at least three independent experiments ($n = 3$), each experiment had a minimum of three replicates of each sample. For comparisons between samples, data was analysed by ANOVA and Tukey's multiple comparison test (SPSS, version 17). Statistical significance was set at $P < 0.01$.

3. Results and discussion

The beetroot juice shot delivers a significant amount of antioxidants in a small, convenient volume as measured by the FRAP assay (697.9 \pm 1.6 $\mu\text{mol}/70$ mL) (Table 2). The TAC of the shot increased significantly ($P < 0.01$) following the in vitro digestion procedure (1740.3 \pm 21.1 $\mu\text{mol}/70$ mL) despite a decrease in TAC between the gastric (2361.2 \pm 20.9 $\mu\text{mol}/70$ mL) and duodenal phases. The differences observed between the gastric and duodenal phase of digestion were in agreement with previous studies (Bermudez-Soto, Tomas-Barberan, & Garcia-Conesa, 2007; Wootton-Beard et al., 2011). In both of these studies the greatest increase in FRAP

was seen following the gastric phase with a marked decrease after the duodenal phase, which may result from biotransformation of antioxidants caused by interaction with the enzymes present in the system (Ryan & Prescott, 2010). The increase in FRAP observed for the beetroot juice shot, following the individual phases of digestion, was proportionally much higher than that of other vegetable juices, most notably the other beetroot juice product produced at the same location, suggesting differences in the processing conditions between the two products. Both beetroot juice products started at relatively similar values prior to digestion and despite significant increases in both, the shot produced a particularly large increase in TAC following both the gastric and duodenal phases, around 2-fold higher than the standard product (Table 2). These results clearly demonstrate that antioxidants contained in this particular product become more accessible following digestion than those in other vegetable juice products.

The in vitro digestion model gives an indication as to the bioaccessibility of vegetable juice antioxidants in a biological system, since the model is designed to simulate in vivo digestion. Bioaccessibility specifically refers to the amount of antioxidants which are potentially presented to the intestinal brush border for absorption. It differs significantly from bioavailability which refers to the amount of antioxidants which pass this intestinal border and are available for use. Furthermore the model omits the colonic phase of the digestive process. Bioaccessibility can however, be used to suggest the rank order of products, and infers that if more antioxidants are presented to the intestinal brush border it is likely that more will be absorbed.

Additionally, Beetroot juice compares favourably with other well accepted high-antioxidant products such as pomegranate and cranberry juices, particularly in terms of bioaccessibility. Research from our laboratory has shown that pomegranate and cranberry juice (both fresh and long life), display FRAP values of 16,789–20,063 $\mu\text{mol}/\text{L}$ and 8419–8570 $\mu\text{mol}/\text{L}$ respectively, after the same in vitro digestion procedure (Ryan & Prescott, 2010). This is significantly lower

280 than the value observed for the beetroot juice shot following
281 digestion ($24,862 \pm 300 \mu\text{mol/L}$).

282 The high TAC of the beetroot juice shot can be largely
283 attributed to its high polyphenol content. The shot was a con-
284 centrated source of dietary polyphenols as measured by the
285 FC method ($68.4 \pm 0.3 \text{ mg GAE/70 mL}$) (Table 2). In similarity
286 with the trends observed in the FRAP assay, the TP content in-
287 creased 5-fold following the gastric phase of the in vitro diges-
288 tion to $341.6 \pm 4.8 \text{ mg GAE/70 mL}$. Following the duodenal
289 phase the TP content was still 3.3-fold higher than the level
290 prior to digestion ($223.2 \pm 5.4 \text{ mg GAE/70 mL}$). The term poly-
291 phenol refers to a particularly wide range of compounds
292 found almost ubiquitously amongst plant foods, in order to
293 be classified as a polyphenol, a compound must simply con-
294 tain one or more aromatic rings and at least two hydroxyl
295 groups according to a recent review by Sies (2010). Despite
296 such a broad classification, Sies (2010) comments that any
297 compound matching these criteria can be considered biolog-
298 ically interesting. With criteria such as these it becomes ever
299 more important to quantify the total polyphenolic content of
300 food products, since all polyphenolic compounds contained
301 in a product may contribute to health. In beetroot, both its
302 pigment and its polyphenol content are related to betanin
303 and indicaxanthin concentrations.

304 Opinions had once focussed on polyphenols primarily as
305 antioxidants with a direct action against RONS, and to a degree
306 this may well be the case. However, due to the relatively low
307 concentrations of free circulating polyphenols found in the
308 body amongst other factors such as biotransformation and a
309 lack of clinical data from intervention trials, researchers have
310 begun to focus on other mechanisms to define the health ben-
311 efits inferred by increasing fruit and vegetable consumption. In
312 recent years polyphenols, have been shown to inhibit alpha-
313 glucosidase/maltase, as well as potentially stimulating insulin
314 secretion which may reduce the absorption of glucose into the
315 blood stream, having profound implications for diabetes man-
316 agement (McDougall & Stewart, 2005). Additionally, they have
317 been shown to display important indirect functions, acting as
318 a 'switch' in redox-dependant signalling pathways and affect-
319 ing the pro-inflammatory effects of RONS through such actions
320 as the modulation of nuclear factor-kappa B activation (Rah-
321 man, Biswas, & Kirkham, 2006).

322 4. Conclusion

323 Given the potential for the multiple health benefits of poly-
324 phenol consumption, beverages containing high levels can
325 be considered a positive addition to the diet. Additionally,
326 both sensory characteristics and convenience would appear
327 to be important factors in obtaining an increased level of con-
328 sumption amongst the general public. Beetroot juice provides
329 a significant source of dietary polyphenols and in particular, a
330 shot can provide a significant quantity of these bioactive com-
331 ponents together with a convenient method for consumption.

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solely for the purpose of providing specific information and
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